

# SERAMPORE GIRLS' COLLEGE

DEPARTMENT OF CHEMISTRY

DEPARTMENTAL REPORT

2024-2025

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## **1.Number of Students**

Year	Minor
SEM 1	00
SEM 2	00
SEM 3	13
SEM 4	13
SEM 5	5
SEM 6	5

Total Number of Students –**18**

## **2.No. of Teachers**

GuestTeacher–**02**

## **3.Student- Teacher ratio**

**9:1**

## 4.Class Routine

Serampore Girls' College

Routine –Chemistry (ODD semester)-2024-25 (July-December, 2024)

Day	SEM	9:45-10:45	10:45-11:45	11:45-12:45	12:45-1:45	2:00-3:00	3:00-4:00
Monday	I						
	III						
	V						
Tuesday	I						
	III				CEMG(MM)(SUB 1) NB 16	CEMG(MM)(SUB 3) NB 19	
	V		CEMG(MM)(P) C LAB	CEMG(MM)(P) C LAB			
Wednesday	I						
	III			CEMG(SD)(SUB 1)SEC NB 19		CEMG(SD)(SUB 3) NB 16	
	V						
Thursday	I						
	III		CEMG(SUB 1)(MM)(P) C LAB	CEMG(SUB 1)(MM)(P) C LAB	CEMG(MM)(SUB 1) NB 16		
	V					CEMG(MM) NB 16	
Friday	I						
	III						
	V						
Saturday	I						
	III			CEMG-SEC (SD)(SUB 3)	CEMG(SD)(SUB 3)(P) CLAB	CEMG(SD)(SUB 3)(P) CLAB	
	V						

SERAMPORE GIRLS' COLLEGE  
Routine –Chemistry (EVEN semester)-2024-25



Day	SEM	9:45-10:45	10:45-11:45	11:45-12:45	12:45-1:45	2:00-3:00	3:00-4:00
Monday	II						
	IV						
	VI						
Tuesday	II						
	IV				CEMG(SUB3)(MM) CLAB	CEMG(SUB3)(MM) CLAB	
	VI			CEMG(MM) NB 16			
Wednesday	II						
	IV				CEMG(SUB1)(SD) C LAB	CEMG(SUB3)(SD) C LAB	
	VI						
Thursday	II						
	IV				CEMG(SUB3)(SD)NB19 CEMG(SUB1)(MM)CLAB	CEMG(SUB1)(MM)CLAB	
	VI		CEMG(P)(MM) CLAB	CEMG(P)(MM) CLAB			
Friday	II						
	IV						
	VI						
Saturday	II						
	IV		CEMG(SUB3)(MM) C LAB	CEMG(SUB1)(MM) NB 19			
	VI	CEMG(MM) NB 16					



## **5.Departmental Activities**

### **A.Teachers' Day Celebration**

**Date:**01/10/2024

**Time:**11:00 a.m.-3:30p.m

**Venue:**NB 21

**Department:** Chemistry, Computer science, Mathematics, Physics, Electronic science, Botany, Zoology

**Objective:** To celebrate Teacher's Day

**Number of attendee:** Student:26 Teacher:8

**Outcome:**It enhanced the teacher student relationship







## B.Educational Tour

Session: **2024-2025**

EventType: **Educational Tour**

EventDate: **20/12/2024**

Attendance: **10**

Objective:

- [1] To enhance knowledge and understanding
- [2] To promote teamwork and social skills
- [3] To encourage independent learning
- [4] To promote creativity and critical thinking
- [5] To provide a break from the monotony of classroom learning

**Serampore Girls' College**  
(Estd. : 1981)  
Accredited by NAAC  
13, T. C. GOSWAMI STREET, SERAMPORE, HOOGHLY, PIN - 712 201, WEST BENGAL.  
Ph: (033) 2652 9334 • (033) 2652 7840 (Principal) • 26623723 (Office) • 2652-0808  
E-mail : science@seramporegirlscollege.org • sgc.scdept@gmail.com

Ref. No. .... Date... 12.12.24.

**Department of Biological Science**

**NOTICE**

Dept. of Biological Science (Botany, Zoology and Chemistry) along with Computer science, Mathematics, Physics and Electronic Science have arranged an Educational Tour for B.Sc. (Major / Minor) semester I and III students on 20<sup>th</sup> December (Friday) to Science City, Kolkata. Interested students have to submit Rs. 450 to Mrs. Nabamita Nath till 17<sup>th</sup> December, 2024. Students are requested to carry Identity Card, small tiffin and sufficient Water. Lunch will be provided. Further information can be gathered from the department.

*Sandhya*  
Principal  
Serampore Girls' College  
Serampore, Hooghly

*Achhallopadhyay* 12/12/24  
Department of Biological Science  
Serampore Girls' College





## 6. Teaching Learning Methods

### Continuous Internal Examination (CIE)

SERAMPORE GIRLS' COLLEGE			
Continuous Internal Evaluation DEC 2024 for CIE for 3rd Semester			
	10:30am-11:30 am	12:30-1:30 pm	2:30-3:30pm
	3rd sem Major DSCC 3 -	3rd sem Major DSCC 4	3rd sem major minor HNE
02-12-2024 Monday	3rd MDC CC Subject 1 History, Education, Sociology, Geography, Music, Philosophy, Botany, Mathematics Chemistry	3rd sem MDC CC Subject 2 English, Bengali, Political Science, Hindi, Urdu, Economics, Physics, Electronics, Zoology	3rd sem MDC minor
03-12-2024 Tuesday			3rd sem Major SEC + 3rd sem MDC CC Physical Education, Sanskrit and Computer Science

*Som 23/11/24*  
 Principal  
 Serampore Girls College  
 Serampore, Hooghly

Ujjam Auluck  
 Convenor, Routine Committee  
 Serampore Girls' College

SERAMPORE GIRLS' COLLEGE			
Continuous Internal Evaluation NOV 2024 for CIE for 1st, and 5th Semester			
	10:30am-11:30 am	12:30-1:30 pm	1:30-2:30pm
	5th sem Hons CC11 and Sem 1 Sanskrit Minor +MDC	5th sem Hons CC12	SEM 1 Minor +(MDC Subject 1) History, Education, Sociology, Geography, Music, Philosophy, Botany, Mathematics
11-11-2024 Monday	5th Sem General Paper DSE - Geography, Sociology, Sanskrit, Hindi, Urdu, Music, Chemistry, Computer Science	5th Sem General DSE - English, Bengali, Political Science, Philosophy, Economics, Physics, Electronics, Botany	
12-11-2024 Tuesday	5th sem Hons DSE A1 and Sem 1 Physical Education Minor+MDC	5th sem Hons DSE B1	SEM 1 Minor + (MDC Subject 2) English, Bengali, Political Science, Hindi, Urdu, Economics, Physics, Electronics,
	5th Sem General DSE - History, Education, Physical Education, Zoology, Mathematics	5th Sem General SEC Paper	

*Som 3/10/24*  
 Principal  
 Serampore Girls College  
 Serampore, Hooghly

Ujjam Auluck 3/10/24  
 Convenor, Routine Committee  
 Serampore Girls' College

# Serampore Girls' College

Continuous Internal Assessment-2024

**CEMG-SEM-V**

**Paper-DSE-A-2**

Full Marks: 20

Answers the Question : (10x2)=20

1. What is Annealing of glass?
2. Why soda lime glass is used?
3. What is glass wool?
4. Write down the chemical reactions involved in the preparation of Pb-azide.
5. Write down the full form of PETN and structure.
6. What is the significance of the fertilizer having grade 14-35-14?
7. Give an example of non-nitrogenous fertilizer with chemical formula?
8. What do you mean by pyrex glass?
9. State two important features of borosilicate glass?
10. Draw the chemical structure of RDX and describe its synthesis?

Full Marks : 20 .....

Madhusree Mal  
Signature of the Examiner



## SERAMPORE GIRLS' COLLEGE

Continuous Internal Evaluation (CIE)

3<sup>rd</sup> Semester

Marks – 20

Paper-MDC-I

Answer any 5 questions. 4x5=20

1. What are extensive and intensive variables?

Clarify the following into extensive and intensive variables –

i) molar volume, (ii) specific heat capacity, (iii) entropy, (iv) chemical potential.

2. Prove that,  $PV^\gamma = \text{constant}$ , for an adiabatic system, where,  $\gamma = \frac{C_p}{C_v}$ .

3. Find  $C_p - C_v$  for any substance and from this show that for an ideal gas,  $C_p - C_v = R$ .

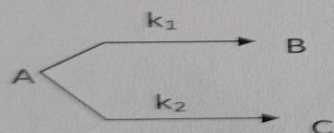
4. State that for an ideal gas adiabatic free expansion must be isothermal.

5. State two features of zero order reaction.

6. Calculate the ratio of  $t_{3/4}$  and  $t_{1/2}$  for a first order reaction. +

7. Short note on consecutive reaction.

8. Give plot of concentration of A, B, C



i)  $\frac{k_1}{k_2} = 2$ , (ii)  $\frac{k_1}{k_2} = \frac{1}{3}$

## SERAMPORE GIRLS' COLLEGE

Continuous Internal Evaluation (CIE)

3<sup>rd</sup> Semester

Marks – 20

Paper-MDC-III

Answer only 5 questions 4x5=20

1. Draw the Newman projection formula for the eclipsed and staggered form of :-

i) 1-bromo -2-chloroethane

ii) 1,2- dibromopropane.

2. What is gauche butane interaction. Mention the stable conformations of 1,2- dihydroxyethane, 1,2- dibromoethane and n-propyl chloride with proper justification.

3. Draw the most stable conformer of  $\text{HOCH}_2\text{CH}_2\text{F}$ . Give reasons.

4. Explain the difference between configuration and conformation. Draw the energy diagram of butane as a function of rotation about  $\text{C}_2\text{-C}_3$  bond and label the maxima and minima with proper configurations.

5. Write down the differences between resonance and tautomerism.

6. The enol content of acetylacetone ( $\text{CH}_3\text{COCH}_2\text{COCH}_3$ ) at equilibrium, is very high (80%), Give reasons.

7. What do you mean by ring- chain and valence tautomerism?

8. Phenol and para-nitro phenol which is stronger acid and why?

9. Write down the correct basicity order in gas phase  $\text{CH}_3\text{NH}_2$ ,  $(\text{CH})_2\text{NH}$ ,  $(\text{CH}_3)_3\text{N}$ .







# Serampore Girls' College

Continuous Internal Assessment-2025

CEMG-SEM-VI

Paper-DSE-B-2

Full Marks: 20

Date: 30/4/2025

Time: 2.30-3.30PM

Answers the Question :(10x2)=20

1. When an organic compound is called optically active?
2. What is Optical Purity (OP) and Enantiomeric Excess (ee)? Give a mathematical expression.
3. What is Molar Absorptivity and write down its unit.
4. Define wavenumber and wavelength also write down the relation between them.
5. What do you mean by Hypsochromic shift and Bathochromic shift, draw the picture.
6. Explain the Lambert-Beer's law, and what its limitation is?
7. What is Chromophores and Auxochromes, give example.
8. A 0.01(M) solution of an unknown compound 'X' shows an absorbance of 0.25, Calculate the value of its molar extinction coefficient, path length 1cm.
9. Describes about Spectrophotometry and Colourimetry.
10. For Sucrose  $C_{12}H_{22}O_{11}$   $[\alpha]_D^{20} = +66.5^\circ$ . Calculate the molecular rotation of sucrose.

